

MolecModels3DTM

Cat.No. 1021000

Contents

Organic Stereochemistry (Student) Set

Qty	Element	colour	holes / type	Dia mm
30	Hydrogen H	white	1molydome®	17
14	Carbon C	black	4 tetrahedral.	23
6	Carbon C	black	5 tribipyramidal	23
6	Oxygen O	red	2 angular	23
4	Nitrogen N	blue	4 tetra.	23
8	Chlorine Cl	green	1 hole.	17
2	Bromine Br	orange	1 hole	17
2	Iodine I	purple	1 hole	17
2	Metal	grey	6 holes	23
	Orbitals			
6	p-orbital	Pink	2D flat	
6	p-orbital	purple	2D flat	
	Bonds			
40	Link -medium	grey	single bonds	19 / 31 *
12	Link -long flexible	grey	double/triple	32 / 43 *
50	Link -short	white	all	2 / 10 *
1	Tool			* total

Isomerism – Molecules with the same molecular formula but different structures.

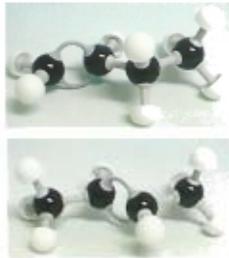
Stereoisomers – Molecules that differ because of the different locations the atoms adopt in space.

Optical isomerism
e.g. Lactic acid



Positional isomerism
e.g. Butene C₄H₈

The position of the non-carbon group, atom or multiple bond varies, giving rise to more than one option.



Skeletal isomerism
e.g. Butane C₄H₁₀

Note the carbon skeleton back bone differences



Functional groups e.g. C₂H₆O These give rise to 2 different functional groups.
In this example an alcohol (ethanol) and an ether are shown.



Geometric isomerism

Where rotation along a bond is not possible, such as in a double bond when a *cis*- or *trans* isomer is possible.



Cis & trans versions of 1,2-Cyclopentanediol
Puckered versions.



Conformational isomerism

Free rotation about the carbon-carbon single bond.
e.g. ethane



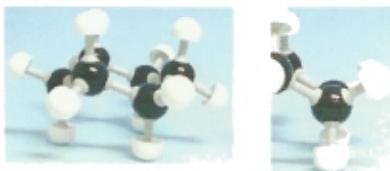
Conformations of Cyclohexane

Boat (perfectly staggered)

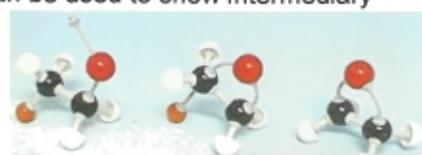
Note the "crowding of the 2 flagpole hydrogens".



Now "flip" the model into the Chair conformation.
Chair (eclipsed)



5-hole carbon atoms can be used to show intermediary stages during the hydrolysis of an epoxide, leading to 1,2-diols and is stereospecific.



Molecular orbital explanation of thermal cyclization of butadienes.

Ground state (homo)



Conrotation of the excited state



Benzene C₆H₆

Molecular orbital model of benzene made using the 5 hole carbons and pink and purple p-orbital lobes.



Disassembly of Compact Models

Please read the following instructions for the recommended use of the Link Remover tool.